Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

### Yan-Jun Li<sup>a,b</sup>\* and Hong-Wu He<sup>b</sup>

<sup>a</sup>College of Chemical Engineering and Technology, Wuhan University of Science and Technology, Wuhan 430081, People's Republic of China, and <sup>b</sup>Key Laboratory of Pesticides and Chemical Biology of the Ministry of Education, College of Chemistry, Central China Normal University, Wuhan 430079, People's Republic of China

Correspondence e-mail: yanwatercn@mail.wust.edu.cn

#### Key indicators

Single-crystal X-ray study T = 272 KMean  $\sigma(\text{C}-\text{C}) = 0.002 \text{ Å}$  R factor = 0.039 wR factor = 0.109 Data-to-parameter ratio = 18.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

## Dimethyl {1-[2-(2-fluorophenoxy)acetoxy]ethyl}phosphonate

The crystal structure of the title compound,  $C_{12}H_{16}FO_6P$ , is stabilized by intermolecular  $C-H\cdots O$  hydrogen bonds.

Received 8 March 2006 Accepted 24 March 2006.

#### Comment

 $\alpha$ -Hydroxyalkylphosphonate derivatives are of considerable interest as potential biologically active compounds or pharmaceuticals (He, 2003). We have synthesized a new  $\alpha$ hydroxyethylphosphonate derivative, (I). Here, we report its structure (Fig. 1).



Selected bond distances and angles are listed in Table 1. The molecules are connected by intermolecular  $C-H\cdots O$  hydrogen bonds (Table 2), forming a layer parallel to the (010) plane (Fig. 2).

#### **Experimental**

The title compound was synthesized by the reaction of dimethyl 1hydroxyethylphosphonate and 2-fluorophenoxyacetyl chloride according to a literature procedure (He *et al.*, 1998). Crystals suitable



© 2006 International Union of Crystallography All rights reserved

# The structure of (I), showing the atom-labeling scheme. Displacement ellipsoids are drawn at the 50% probability level.

## organic papers

for X-ray diffraction were obtained by slow evaporation of a petroleum ether/dichloromethane (4:1  $\nu/\nu$ ) solution at 293 K.

 $D_x = 1.401 \text{ Mg m}^{-3}$ 

Mo  $K\alpha$  radiation Cell parameters from 4456

reflections

T = 272 (2) K

Block, colorless

 $0.30 \times 0.30 \times 0.20 \text{ mm}$ 

 $w = 1/[\sigma^2(F_0^2) + (0.0569P)^2]$ 

where  $P = (F_0^2 + 2F_c^2)/3$ 

+ 0.3545P]

 $(\Delta/\sigma)_{\rm max} < 0.001$ 

 $\Delta \rho_{\rm max} = 0.33 \text{ e} \text{ Å}^{-3}$ 

 $\Delta \rho_{\rm min} = -0.28 \text{ e } \text{\AA}^{-3}$ 

 $\begin{array}{l} \theta = 2.2 {-} 28.2^{\circ} \\ \mu = 0.22 \ \mathrm{mm}^{-1} \end{array}$ 

#### Crystal data

 $\begin{array}{l} C_{12}H_{16}FO_6P\\ M_r = 306.22\\ Monoclinic, P2_1/c\\ a = 8.7321 \ (8) \ \AA\\ b = 18.3099 \ (16) \ \AA\\ c = 9.2048 \ (8) \ \AA\\ \beta = 99.448 \ (2)^\circ\\ V = 1451.7 \ (2) \ \AA^3\\ Z = 4 \end{array}$ 

#### Data collection

Bruker SMART CCD area-detector<br/>diffractometer2851 reflections with  $I > 2\sigma(I)$ <br/> $R_{int} = 0.018$ <br/> $\theta_{max} = 27.5^{\circ}$ <br/>Absorption correction: none<br/>9060 measured reflections $\theta_{max} = 27.5^{\circ}$ <br/> $h = -11 \rightarrow 10$ <br/> $k = -23 \rightarrow 23$ <br/>314 independent reflections

#### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.039$   $wR(F^2) = 0.109$  S = 1.033314 reflections 184 parameters H-atom parameters constrained

#### Table 1

Selected geometric parameters (Å, °).

C1-O1	1.3704 (19)	C9-P1	1.8110 (15)
C7-O1	1.4178 (18)	C11-O4	1.447 (2)
C8-O2	1.1927 (17)	O4-P1	1.5682 (13)
C8-O3	1.3454 (17)	O5-P1	1.4649 (11)
C9-O3	1.4573 (16)	O6-P1	1.5688 (13)
O1-C1-C6	126.52 (14)	C8-O3-C9	117.31 (11)
O3-C8-C7	108.99 (12)	O5-P1-O6	116.09 (7)
C1-O1-C7	116.99 (12)	O4-P1-O6	102.03 (8)

#### Table 2

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
C3-H3···O5 <sup>i</sup>	0.93	2.51	3.416 (2)	165
$C7-H7A\cdots O5^{ii}$	0.97	2.39	3.2793 (19)	152
С9−Н9…О2	0.98	2.33	2.7241 (19)	103

Symmetry codes: (i) x + 1, y, z + 1; (ii) x + 1, y, z.



#### Figure 2

The packing of (I), showing  $C-H\cdots O$  hydrogen bonds as dashed lines. Suffixes a and b correspond to symmetry codes (i) and (ii), respectively, in Table 2.

H atoms were placed at calculated positions (C-H = 0.93–0.98 Å) and refined as riding, with  $U_{iso}(H) = 1.2$  or  $1.5U_{eq}(C)$ .

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2001); software used to prepare material for publication: *SHELXTL*.

The authors acknowledge financial support from the National Natural Science Foundation of China (No. 20072008).

#### References

Bruker (1997). SMART (Version 5.054). Bruker AXS Inc., Madison, Wisconsin, USA.

Bruker (1999). SAINT (Version 6.01). Bruker AXS Inc., Madison, Wisconsin, USA.

Bruker (2001). SHELXTL (Version 6.12). Bruker AXS Inc., Madison, Wisconsin, USA.

He, H.-W. (1998). Chin. Chem. Lett. 9, 415-416.

He, H.-W. (2003). Chin. J. Org. Chem. 23, 155-161.

Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.